**Conclusion:**

In the chart below, write a BALANCED equation for each of the reactions that took place.

(You may balance on a separate sheet of paper first to make your chart neat and legible.)

Do not write an equation for no reactions (NR).

|  |
| --- |
| **Complete Balanced Equations** |
| Zn |
| Zn |
| Zn |
| Pb |
| Pb |
| Pb |
| Cu |
| Cu |
| Cu |
| Al |
| Al |
| Al |
| Mg |
| Mg |
| Mg |

What did you learn about single replacement reactions?